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**CHE 532/596 Course Syllabus**

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**CHE 532/596 – Core ChE Concepts: II**

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**Section 651****Summer 2024 (10-week session)****3 Credit Hours**

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**Course Description**

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This online chemical engineering bridging course is the second part of a comprehensive two-course sequence (three credits each) consisting of core undergraduate level chemical engineering topics. This homework-intensive course is designed to prepare students with backgrounds in chemistry, biology and other non-chemical engineering subjects for graduate study in chemical engineering. It is not equivalent to a four-year degree in chemical engineering, but the proposed course is designed to help you obtain the additional knowledge and skills to successfully undertake graduate classes offered by a chemical engineering program.

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**Learning Outcomes**

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By the end of this course, students should be able to:

- Describe and analyze energy and heat transport through single and multiple phases as well as steady-state and transient mass transfer processes.
- Apply first and second laws to multicomponent systems to calculate thermodynamic properties.
- Develop rate equations from known elementary reactions or available kinetic data, size a reactor given reaction kinetics, and evaluate reactor design options for multiple reactions.

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**Course Structure**

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Students will watch the short lecture videos for each module, as well as videos that detail worked out examples. Students will submit homework individually, and after the late due date may access annotated on-line solutions. There will be three tests and one final exam.

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**Instructor**

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**Dr. Hassan Golpour - Instructor****Email:** [hgolpou@ncsu.edu](mailto:hgolpou@ncsu.edu)**Phone:** 919-513-0652**Office Location:** 2086-A Engineering Building I (EB1)

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**Course Meetings**

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None.

## Course Materials

### Textbooks

*This textbook, also used in Course I, is required.*

**Introduction to Chemical Engineering Thermodynamics** - J.M. Smith, Hendrick Van Ness, Michael Abbott, and Mark Swihart

**Edition:** 8th edition

**ISBN:** 978-1259696527

**Web Link:** <https://www.mheducation.com/highered/product/introduction-chemical-engineering-thermodynamics-smith-van-ness/M1259696529.html>

**Cost:** \$160.28 (hard copy)

*This textbook, also used in Course I, is required.*

**Transport Phenomena** - R. Bryon Bird, Warren E. Stewart, Edwin N. Lightfoot

**Edition:** Revised 2nd edition

**ISBN:** 978-0470115398

**Web Link:** <https://www.wiley.com/en-us/Transport+Phenomena%2C+Revised+2nd+Edition-p-9780470115398>

**Cost:** \$94.99 (hard copy)

**Elements of Chemical Reaction Engineering** – H. Scott Fogler

**Edition:** 5th edition

**ISBN:** 978-0133887518

**Web Link:** <https://www.pearson.com/us/higher-education/program/Fogler-Elements-of-Chemical-Reaction-Engineering-5th-Edition/PGM265957.html>

**Cost:** \$150.46 (hard copy)

*This textbook is required.*

## Expenses

None.

## Materials

None.

## Requisites and Restrictions

### Prerequisites

There are no formal (i.e. no course number) prerequisites for the course. However, it is recommended that students have taken general chemistry as well as math courses up to and including differential equations.

### Co-requisites

None.

**Restrictions**

None.

**General Education Program (GEP) Information****GEP Category**

This course does not fulfill a General Education Program category.

**GEP Co-requisites**

This course does not fulfill a General Education Program co-requisite.

**Transportation**

This course will not require students to provide their own transportation. Non-scheduled class time for field trips or out-of-class activities is NOT required for this class.

**Safety & Risk Assumptions**

None.

**Grading****Grade Components**

Component	Weight
Homework Sets	10%
Quizzes	5%
3 Tests	20% each
Final Exam	25%

**Letter Grades**

This course is a graduate course, and as such letter grades adhere to a different standard of expectations than undergraduate courses. With this in mind this course does not necessarily use the standard NCSU letter grading scale.

**Requirements for Credit-Only (S/U) Grading**

Performance in research, seminar and independent study types of courses (6xx and 8xx) is evaluated as either "S" (Satisfactory) or "U" (Unsatisfactory), and these grades are not used in computing the grade point average. For credit only courses (S/U) the requirements necessary to obtain the grade of "S" must be clearly outlined.

## Requirements for Auditors (AU)

Information about and requirements for auditing a course can be found at <http://policies.ncsu.edu/regulation/reg-02-20-04>.

## Policies on Incomplete Grades

If an extended deadline is not authorized by the Graduate School, an unfinished incomplete grade will automatically change to an F after either (a) the end of the next regular semester in which the student is enrolled (not including summer sessions), or (b) by the end of 12 months if the student is not enrolled, whichever is shorter. Incompletes that change to F will count as an attempted course on transcripts. The burden of fulfilling an incomplete grade is the responsibility of the student. The university policy on incomplete grades is located at <http://policies.ncsu.edu/regulation/reg-02-50-03>. Additional information relative to incomplete grades for graduate students can be found in the Graduate Administrative Handbook in Section 3.18.F at [http://www.fis.ncsu.edu/grad\\_publicns/handbook/](http://www.fis.ncsu.edu/grad_publicns/handbook/)

## Late Assignments

Homework is typically due at 11:59 PM ET on the Wednesday of the week the assignment is stated to be due in the course schedule below. Students who are unable to submit the homework on time should contact the instructor in advance of the deadline if possible to explain their circumstances. Unexcused late homework will be accepted with a penalty of -10% for each 24 hours of delay. Homework submitted more than 72 hours late (11:59 PM on Saturday for an 11:59 PM Wednesday due date) will not be accepted. Homework solutions will be accessible on Moodle 72 hours after the assignment is due.

Quizzes are typically due at 11:59 PM ET on the Thursday of the week the assignment is stated to be due in the course schedule below. Late quiz submissions will not be accepted. With this strict penalty in mind, students should be sure to submit quiz solutions well before the deadline to allow time for any unforeseen technical problems.

Each test/exam will be given as a “take-home” exam that will be posted to the Moodle site. Typically, exams will be posted at 8:00 AM ET on the first day of the testing window, and will remain there until 11:59 PM ET on the second day of the testing window, i.e. a ~40 hour testing window is provided. For the final exam, the exam will be posted at 8:00 AM on Monday of finals week and will remain there until 11:59 PM ET on Tuesday of finals week. A submission link similar to that used for homework will appear on the Moodle site as well – for regular exams, students must submit all solutions (as one PDF file) by 11:59 PM ET on the second day of the testing week; this deadline is 11:59 PM ET on Tuesday of finals week for the final exam. Late submissions of any exam are immediately penalized 10% (AKA 10 points for an exam graded out of 100 points). Each pro-rated hour of delay after the corresponding deadline will decrease the grade by an additional 10%; for example, if an exam is submitted 15

minutes late, a 12.5% deduction will be applied – in other words, the immediate 10% deduction, as well as an additional 2.5% deduction for 0.25 hours late at 10% per hour. With these penalties in mind, students should be sure to submit solutions well before the deadline to allow time for any unforeseen technical problems.

### **Attendance Policy**

For complete attendance and excused absence policies, please see <http://policies.ncsu.edu/regulation/reg-02-20-03>

### **Attendance Policy**

None.

### **Absences Policy**

None.

### **Makeup Work Policy**

None.

### **Additional Excuses Policy**

None.

### **Academic Integrity**

#### **Academic Integrity**

Students are required to comply with the university policy on academic integrity found in the Code of Student Conduct found at <http://policies.ncsu.edu/policy/pol-11-35-01>

It is the instructor's understanding and expectation that the student's name or signature on any test or assignment means that the student contributed to the assignment in question (if a group assignment) and that they neither gave nor received unauthorized aid (if an individual assignment). Authorized aid on an individual assignment includes discussing the interpretation of the problem statement, sharing ideas or approaches for solving the problem, and explaining concepts involved in the problem. Any other aid would be unauthorized and a violation of the academic integrity policy. Any computer work submitted must be completed on your own personal computer or from your own NCSU account to avoid confusion about the origin of the file, and no sharing of files in any way is allowed. All cases of academic misconduct will be submitted to the Office of Student Conduct. Students found guilty of academic misconduct will be subject to, at a minimum, a zero on the assignment in question, up to a zero for that course component (e.g. a zero for the homework portion of the final grade), or a failing grade in the course, depending on the nature of the violation.

## **Academic Honesty**

See <http://policies.ncsu.edu/policy/pol-11-35-01> for a detailed explanation of academic honesty. Authorized aid on an individual assignment includes discussing the interpretation of the problem statement, sharing ideas or approaches for solving the problem, and explaining concepts involved in the problem. Any other aid would be unauthorized and a violation of the academic integrity policy. Coordinating with other students, or individuals outside the course, on any assignments and/or exams in any way can result in a 0 on the final exam and an academic integrity violation with the university. Unauthorized aid additionally includes uploading or accessing on-line solutions (e.g. Chegg), whether that involves posted problem solutions or paying someone to assist in solving problems. Posting questions from the exam elsewhere (such as homework help websites) can result in a 0 on the final exam and an academic integrity violation with the university.

All cases of academic misconduct will be submitted to the Office of Student Conduct. Students found guilty of academic misconduct will be subject to, at a minimum, a zero on the assignment in question, up to a zero for that course component (e.g. a zero for the homework portion of the final grade), or a failing grade in the course, depending on the nature of the violation. In addition, if you are found guilty of academic misconduct in the course, you will be on academic integrity probation for the remainder of your years at NCSU and may be required to report your violation on future professional school applications.

## **Honor Pledge**

It is the instructor's understanding and expectation that the student's name / signature on any test or assignment cover sheet means that the student contributed to the assignment in question (if a group assignment) and that they neither gave nor received unauthorized aid (if an individual assignment).

## **Electronically-Hosted Course Components**

Students may be required to disclose personally identifiable information to other students in the course, via electronic tools like email or web-postings, where relevant to the course. Examples include online discussions of class topics, and posting of student coursework. All students are expected to respect the privacy of each other by not sharing or using such information outside the course.

## **Accommodations for Disabilities**

Reasonable accommodations will be made for students with verifiable disabilities. In order to take advantage of available accommodations, students must register with the [\*\*Disability Resource Office\*\*](#) at Holmes Hall, Suite 304, Campus Box 7509, 919-515-7653. For more information on NC State's policy on working with students with disabilities, please see the [\*\*Academic Accommodations for Students with Disabilities Regulation \(REG02.20.01\)\*\*](#). Students needing special accommodations for testing are responsible for

making arrangements to take exams at the on-campus testing center and ensuring that sufficient time is available to complete the exam.

### **Non-Discrimination Policy**

NC State University provides equality of opportunity in education and employment for all students and employees. Accordingly, NC State affirms its commitment to maintain a work environment for all employees and an academic environment for all students that is free from all forms of discrimination. Discrimination based on race, color, religion, creed, sex, national origin, age, disability, veteran status, or sexual orientation is a violation of state and federal law and/or NC State University policy and will not be tolerated. Harassment of any person (either in the form of quid pro quo or creation of a hostile environment) based on race, color, religion, creed, sex, national origin, age, disability, veteran status, or sexual orientation also is a violation of state and federal law and/or NC State University policy and will not be tolerated. Retaliation against any person who complains about discrimination is also prohibited. NC State's policies and regulations covering discrimination, harassment, and retaliation may be accessed at <http://policies.ncsu.edu/policy/pol-04-25-05> or [http://www.ncsu.edu/equal\\_op/](http://www.ncsu.edu/equal_op/). Any person who feels that he or she has been the subject of prohibited discrimination, harassment, or retaliation should contact the Office for Equal Opportunity (OEO) at 919-515-3148.

### **Trans-Inclusive Statement**

In an effort to affirm and respect the identities of transgender students in the classroom and beyond, please contact the instructor if you wish to be referred to using a name and/or pronouns other than what is listed in the student directory.

### **Basic Needs Security**

Any student who faces challenges securing their food or housing or has other severe adverse experiences and believes this may affect their performance in the course is encouraged to notify the professor if you are comfortable in doing so. Alternatively, you can contact the Division of Academic and Student Affairs to learn more about the Pack Essentials program <https://dasa.ncsu.edu/pack-essentials/>

## Course Schedule

**NOTE:** The course schedule is subject to change.

### Week 1 — 5/15 - 5/17

12.1 Fourier's Law of Heat Conduction (Molecular Energy Transport) [BSL 9.1]
12.2 Temperature and Pressure Dependence of Thermal Conductivity [BSL 9.2]
12.3 Heat Transport by Convection: Forced and Free [BSL 4.4, 10.8 & 10.9]
12.4 Combined Mechanisms of Heat Transfer [BSL 10.6]
12.5 Heat Transport by Radiation [BSL 16.4]
13.1a The Energy Equation [BSL 11.1 & 11.2]
13.1b Special Forms of the Differential Energy Equation [BSL 11.1]
13.1c Application of the Energy Equation to Heat Generation in a Cylinder
13.2 Important Dimensionless Groups [BSL 9.1, 10.8, 10.9 & 11.5]
13.3a Heat-Transfer Equipment
13.3b Double Pipe Heat-Exchanger Analysis [BSL 15.4]

### Week 2 — 5/20 - 5/24 – Homework problems in Topics 12 and 13 (5 problems total) due by 11:59 PM on 5/22 and Quizzes 1 and 2 due by 11:59 PM on 5/23

14.1 Fick's Law of Binary Diffusion (Molecular Mass Transport) [BSL 17.1]
14.2 Temperature and Pressure Dependence of Diffusivities [BSL 17.2]
14.3 Mass Transfer Notation and Flux Definitions [BSL 17.7 & 17.8]
15.1 Equation of Continuity for Mass Transfer [BSL 19.1]
15.2 Special Forms of the Equation of Continuity [BSL 19.1]
15.3 Mass Transfer Equipment [BSL 22.8]

### Week 3 — 5/27 - 5/31 – Homework problems in Topics 14 and 15 (4 problems total) due by 11:59 PM on 5/29 and Quizzes 3 and 4 due by 11:59 PM on 5/30

No new material this week – use this time to take a breath and prepare for Test 1!
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### Week 4 — 6/3 - 6/7 – TEST 1 on Topics 12-15 (6/3 – 6/4)

16.1a Fundamental Relation and Chemical Potential [SVAS 10.1 & 10.2]
16.1b Chemical Equilibrium [SVAS 10.2]
16.2 Partial Properties [SVAS 10.3]
16.3 The Ideal-Gas-State Mixture Model [10.4]

16.4 Fugacity and Fugacity Coefficient: Pure Species [SVAS 10.5]
16.5 Fugacity and Fugacity Coefficient: Species in Solution [SVAS 10.6]
16.6 Generalized Correlation For Fugacity Coefficient [SVAS 10.7]
16.7 The Ideal Solution Model [SVAS 10.8]
16.8 Excess Properties [SVAS 10.9]
17.1 Property Changes of Mixing [SVAS 11.1]

**Week 5 — 6/10 - 6/14 – Homework problems in Topics 16 and 17 (6 problems total) due by 11:59 PM on 6/12 and Quiz 5 due by 11:59 PM on 6/13**

18.1 The Phase Rule - Duhem's Theory [SVAS 12.2]
18.2 VLE: Qualitative Behavior – The PTxy Diagram [SVAS 12.3]
18.3 VLE: Qualitative Behavior – PT, Pxy, and Txy Diagrams [SVAS 12.3]
18.4 VLE: Qualitative Behavior – Critical Point/Retrograde Condensation [SVAS 12.3]
18.5 VLE: Qualitative Behavior – Azeotropes [SVAS 12.3]
19.1 Excess Gibbs Energy and Activity Coefficients [SVAS 13.1]
19.2 The Gamma/Phi Formulation of VLE [SVAS 13.2]
19.3 Modified Raoult's Law, & Henry's Law [SVAS 13.3]
19.4 Flash Calculations [SVAS 13.8]

**Week 6 — 6/17 - 6/21 – Homework problems in Topics 18 and 19 (6 problems total) due by 11:59 PM on 6/19 and Quizzes 6 and 7 due by 11:59 PM on 6/20**

No new material this week – use this time to take a breath and prepare for Test 2!
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**Week 7 — 6/24 – 6/28 – TEST 2 on Topics 16-19 (6/24 – 6/25)**

20.1 The Rate of Reaction [FOG 1.1]
20.2 The General Mole Balance Equation & Batch Reactors [FOG 1.2 & 1.3]
20.3 Continuous-Flow Reactors(CSTR, Tubular, PBR) [FOG 1.4]
21.1 Definition of Conversion & Batch Reactor Design Equation [FOG 2.1 & 2.2]
21.2 Design Equations For Flow Reactors (CSTR, Tubular, Packed Bed) [FOG 2.3]
21.3 Reactors in Series (CSTR, PFR, Combination) [FOG 2.5]

**Week 8 — 7/1 - 7/5 – Homework problems in Topics 20 and 21 (4 problems total) due by 11:59 PM on 7/3 (no quiz this week)**

22.1 Basic Definitions, The Reaction Order and Rate Law [FOG 3.1 & 3.2]
22.2 Non-Elementary Rate Laws [FOG 3.2]

22.3 Reaction Rate Constant, Reactor Sizing And Design Summary [FOG 3.3 & 3.4]
23.1 Batch Systems [FOG 4.1]
23.2a Flow Systems [FOG 4.2]
23.2b Flow Systems for Gases [FOG 4.2]

**Week 9 — 7/8 - 7/12 – Homework problems in Topics 22 and 23 (2 problems total) due by 11:59 PM on 7/10 and Quizzes 8 and 9 due by 11:59 PM on 7/11**

24.1 Design Structure For Isothermal Reactors [FOG 5.1]
24.2 Batch Reactors [FOG 5.2]
24.3 Single CSTR [FOG 5.3]
24.4 CSTR in Series [FOG 5.3]
25.1 Mole Balances on CSTR, PFR, PBR and Batch Reactors [FOG 6.2]

**Week 10 — 7/15 - 7/19 – Homework problems in Topic 24 (3 problems total) due by 11:59 PM on 7/17 and Quiz 10 due by 11:59 PM on 7/18**

No new material this week – use this time to take a breath and prepare for Test 3!
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**Week 11 — 7/22 - 7/26 – TEST 3 on Topics 20-25 (7/22 – 7/23)**

No new material this week – remember to prepare for next week’s final exam!
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### **Final Exam: July 29-30**

#### **Additional Policies**

1. Each test/exam will be given as a “take-home” exam that will be posted to the Moodle site. Typically, exams will be posted at 8:00 AM ET on the first day of the testing window, and will remain there until 11:59 PM ET on the second day of the testing window, i.e. a ~40 hour testing window is provided. For the final exam, the exam will be posted at 8:00 AM on Monday of finals week and will remain there until 11:59 PM ET on Tuesday of finals week. A submission link similar to that used for homework will appear on the Moodle site as well – for regular exams, students must submit all solutions (as one PDF file) by 11:59 PM ET on the second day of the testing week; this deadline is 11:59 PM ET on Tuesday of finals week for the final exam. Late submissions of any exam are immediately penalized 10% (AKA 10 points for an exam graded out of 100 points). Each pro-rated hour of delay after the corresponding deadline will decrease the grade by an additional 10%; for example, if an exam is submitted 15 minutes late, a 12.5% deduction will be applied – in other words, the immediate 10% deduction, as well as an additional 2.5% deduction for 0.25 hours late at 10% per hour. With these penalties in mind, students should be sure to submit solutions well before the deadline to allow time for any unforeseen technical problems.

2. Student materials (e.g. homework and exams) will be submitted for grading electronically by

submitting a scanned copy of your work in PDF format via the upload link provided for the problem/assignment on the course Moodle site. Submissions must be high-quality scans (not cell phone pictures) in order to ensure legibility for grading. Be aware that typical “scanning” apps that come with iPhones, etc. are often not suitable. It is suggested that students download and use the free CamScanner app for their submissions; other options are used by students at their own risk of point deductions. Since PDF images from computational software (e.g. Excel, MATLAB) typically do not explicitly show all steps of showing work, no images from computational software will be considered in grading of homework or exams.

3. Typically, in the Core ChE Concepts I and II courses, though exams are intended to last for 75 minutes (for midterm exams) or 150 minutes (for final exams), there is no 75 or 150 minute timer; rather, all exams have an extended window (e.g. 40 hours) for submission in an effort to be sensitive/understanding of students who may be juggling work, family, and other responsibilities. With this in mind, DRO has advised the instructor that all students receive the same 40-hour window to complete the exam even if a student’s DRO accommodations allow extended time (e.g. 1.5x or 2x) since extended time is not appropriate considering the intended test duration and the 40 hour testing window. If you would like to use a reduced-distraction testing environment as listed on your accommodation letter, please work with DRO to secure a spot at their testing center. If a student feels any part of this policy does not apply given their specific accommodations through DRO, they should contact the course instructor.

4. In the event a student misses one midterm exam (i.e. Tests 1-3) for any reason, they have the option to take a comprehensive make-up exam in its place at the end of the semester, after the last week of classes. Make-up midterm exams submitted by students who missed the test for a university-approved event are eligible for full credit. Make-up exams submitted by students who missed an exam for a non-university-approved event (e.g. flat tire, overslept, etc.) will be penalized 30 points. In the event a student misses multiple exams during the semester, make-up exams will only be offered for missed exams for which the student has a university-approved excuse – with this in mind, if a student misses multiple exams, any exams a student misses without a university-approved excuse will be assigned a grade of zero. To be clear, this make-up policy does not apply to the final examination – students who miss the final exam should expect to receive a grade of zero on the final exam.

5. Exams are open-book (*hard copy*) and open-notes. You may use a device (computer, tablet, etc.) during the exam only to access materials posted to the course website. Do not access any other websites for assistance during the exam, including YouTube, tutoring websites such as Chegg, AI tools such as ChatGPT, etc.

6. Quiz questions are graded in the following way: if you input the correct answer to a quiz question on your first try, you will receive full credit for the problem; if you are incorrect on your first try on a problem, you will then be given feedback for your second try. If you get the question right on your second try, you will be given 50% of the score for that problem; if you are incorrect on your second try, you will receive zero credit for the problem.

7. Homework is an important element of this class. Most students learn the topics covered in the course by practice and not simply by listening to the lecture or reading the text. Homework

will typically be due via upload to the course Moodle site on Wednesday at 11:59 PM ET. A word of advice --- do not wait until the evening before the assignment is due to start the problems.

8. Homework/Topic Forum Policy: If you have a general question about course topics or homework problems, before sending an email to the course instructor(s) you should post your question on the corresponding topic forum on Moodle. There is a different forum available for each main course topic. Students are encouraged to participate in responding to posted questions on the homework forum.

- Bonus: the student with the most engagement on Moodle (based on answering posted questions on HW forums) will receive Extra Credit (up to 1% of the final course grade).
- Dos: You can post a general question about how to approach a specific problem or, for example, how to find physical properties of specific molecules. You can also ask questions to better understand what a specific problem is asking for.
- Don'ts: Do not post your intermediate or final answers to check with others. Do not ask/answer questions about what specific equations to use for a specific problem

9. Solutions to the homework assignments are expected to be posted on the course website 72 hours after it is due.

10. If you have questions about homework grading, you can access the PDF and video of the worked solution after the late submission deadline (typically Saturday at 11:59 PM for homework due on Wednesday at 11:59 PM). Please review the solution first if you have questions about grading. If, after reviewing the solution, you believe that an error was made in grading the homework (or test question, etc.), email Dr. Golpour with a brief justification/explanation of your request. The “statute of limitations” for submitting a re-grade request is one week after the homework/test/etc. solution is posted.

11. You are not permitted to consult homework solutions from past offerings of this class or to discuss solutions with students who have previously taken it; however, discussions with your classmates (current registered online students) about how to approach problems or about course concepts are encouraged. See the section on academic integrity that describes the boundaries around what is appropriate for consulting with classmates.

12. As specified above, questions regarding course material should be posted to the relevant discussion forum. However, if a personal circumstance arises that you would like to discuss with the instructor, e-mail the instructor to discuss / schedule an appointment to discuss over Zoom.

13. In order to maintain equal avenues of contact for all students regardless of their location, the instructors will not meet in person or by phone to hold office hours or discuss course material.

14. Online class evaluations will be available beginning on or about the last week of class. You will receive an email message from the University directing you to a website where you can login using your Unity ID and complete the evaluations. All evaluations are confidential; instructors never know how any one student responded to any question.

- Evaluation website: <https://classeval.ncsu.edu>

- Student help desk: [classeval@ncsu.edu](mailto:classeval@ncsu.edu)
- More information about ClassEval: <http://oirp.ncsu.edu/eval/clev>

15. All materials in this course (including but not limited to: electronic and hard copies of lecture notes, lecture videos, homework problems and/or solutions, quizzes and/or solutions, exams and/or solutions, or handouts) – made available to you are copyrighted by Dr. Matthew Cooper, Dr. Lisa Bullard and/or Dr. Hassan Golpour and intended only for your personal use. You are not allowed to share content of the class with any person not signed up for the course; a personal, public, or commercial website; or any news or advertising media. To be clear, you are not permitted to upload any content from this course to the web in any form, including but not limited to Chegg, Course Hero, Coursera, Google Drive, etc. If you post course content, you risk violating the instructor’s intellectual property rights as well as NCSU’s Code of Student Conduct. All cases of academic misconduct will be submitted to the Office of Student Conduct. Students found guilty of academic misconduct will be subject to, at a minimum, a zero on the assignment in question, up to a failing grade in the course, depending on the nature of the violation. In addition, if you are found guilty of academic misconduct in the course, you will be on academic integrity probation for the remainder of your years at NCSU and may be required to report your violation on future professional school applications.

16. You may not consult other classmates, individuals, websites (other than the course Moodle site), entities, etc. while completing exams. Do not reveal or discuss the contents of an exam with other parties until after midnight on the day following the two-day testing window for the exam. Accessing any other websites (but particularly “tutoring” websites such as Chegg, Course Hero, etc.) to assist your work on the exam is strictly forbidden. Dr. Golpour is working closely with tutoring websites to identify copyright violations from materials uploaded to them and refer offending cases to NCSU’s Office of Student Conduct in cooperation with the tutoring websites. A violation of the Code of Student Conduct could endanger your graduate career / degree at NCSU. Not only is cheating using tutoring websites unethical, it is simply not worth the downside of getting caught!

17. Artificial intelligence (AI) language tools, such as (but not limited to) ChatGPT, should not be used for course assignments except as explicitly authorized by the instructor. This includes (but is not limited to) the following actions:

- Submitting all or any part of an assignment statement to an AI tool
- Incorporating any part of an AI-generated response in an assignment
- Using AI to brainstorm, formulate arguments, or template ideas for assignments